

## Experiment #12: Acid-Base Titration

NAME: \_\_\_\_\_

Section: \_\_\_\_\_

Date: \_\_\_\_\_

### DATA & LAB REPORT:

Part A: Determining Molarity of NaOH Solution

\*Show ALL Calculation Setups

	Trial 1	Trial 2	Trial 3
Mass of KHP	0.5858 g	0.9510g	
Final Buret Reading	31.10mL	44.10 mL	
Initial Buret Reading	0.73 mL	0.00 mL	
Volume of NaOH Used	30.37 mL	44.10 mL	

1. Calculate moles of KHP used in each trial.

Moles of KHP

Trial 1	Trial 2	Trial 3
_____	_____	_____

2. How many moles of NaOH was used in each trial?

Moles of NaOH

Trial 1	Trial 2	Trial 3
_____	_____	_____

## Lab 39 Acid Base Titration Answers

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**Chemistry in the Laboratory** James M. Postma, Julian L. Robert, J. Leland Hollenberg, 2004-03-12 This clearly written class tested manual has long given students hands on experience covering all the essential topics in general chemistry Stand alone experiments provide all the background introduction necessary to work with any general chemistry text This revised edition offers new experiments and expanded information on applications to real world situations *BIS Exam*

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**Learning**, 1984 **Addison-Wesley Small-scale Chemistry** Dennis D. Staley, Edward L. Waterman, 1995 *Glencoe Algebra 1*, 2001 Government Reports Announcements & Index, 1979 *Monthly List of Russian Accessions* Library of

Congress. Processing Department, 1966-02 **Monthly Index of Russian Accessions** Library of Congress. Processing Department, 1966 *INIS Atomindex*, 1978 **Ceramic Abstracts**, 1954 **Agrindex**, 1994 **Acid-base Titrations**

**in Nonaqueous Solvents** James Sherwood Fritz, 1952 **The Effect of Adding Guided-inquiry to Laboratory Activities in an Acid Base Unit in a High School Chemistry Classroom** Kendon Douglas Smith, 2010 **General Theory of Acid-base Titration** Pat Richard Pondy, 1949

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