

**Ionic Salts Simulation Lab****Learning Goals:**

- Understand how an ionic salt dissolves describing a saturated solution microscopically and macroscopically.
- Calculate solubility in grams/100ml
- Determine why some salts are more soluble than others
- Design and carry out a virtual lab to test a hypothesis

Directions: Go to

[http://phet.colorado.edu/simulations/sims.php?sim=Salts\\_and\\_Solubility](http://phet.colorado.edu/simulations/sims.php?sim=Salts_and_Solubility) and run the Salts and Solubility simulation. Take a few minutes to get use to how the simulation works. Then answer the questions below.

Questions:

- 1) Describe what you observe when you shake table salt into water. Test numbers of particles with different volumes.
  - a. When the salt dissolves, which element becomes the anion (the negatively charged atom) and which one becomes the cation (the positively charged atom)?
  - b. What is the relationship between amount of particles and amount of water? (direct or indirect).
  - c. At  $5.00 \times 10^{23}$  what is the point of saturation (point where you can't dissolve any more salty)?
- 2) Fill in the chart below to see if the ratio of particles to water is constant at the saturation point.

Particles	Water Volume	Ratio (particles/volume)

- a. Calculate the constant below (show your work)

- 3) Test a variety of slightly soluble salts, noting the relationship between numbers of each ion pair. Fill in the chart below to help you see a relationship.
  - a. How many particles of each element need to be present before the salt stops dissolving (point of saturation)?

# Lab 30 Solubility Of A Salt Answers

**Frank Brescia**

## Lab 30 Solubility Of A Salt Answers:

**Laboratory Guide to Pharmaceutical Technique** University of Wisconsin. Department of Pharmacy, 1926

**Chemistry in the Laboratory** James M. Postma, Julian L. Roberts, J. Leland Hollenberg, 2009-08-07 For nearly 40 years Chemistry in the Laboratory has been meeting the needs of teachers and students. This new edition builds on that legacy while addressing cutting edge trends in the chemistry laboratory including forensic chemistry and environmental and green chemistry. As always the new edition of Chemistry in the Laboratory offers precise, easy to follow instructions, helpful illustrations and an emphasis throughout on laboratory safety. Again throughout a Consider This feature encourages users to expand the principles of the experiment into interesting applications, open ended experiments or unexplored corners. Most experiments in the manual can be completed in one lab session but some can be linked or extended for a multi lab project.

**Self-assessment Questions for Clinical Molecular Genetics** Haiying Meng, 2019-05-28 Review Questions of Clinical Molecular Genetics presents a comprehensive study guide for the board and certificate exams presented by the American College of Medical Genetics and Genomics ACMG and the American Board of Medical Genetics and Genomics ABMGG. It provides residents and fellows in genetics and genomics with over 1 000 concise questions ranging from topics in cystic fibrosis to genetic counseling to trinucleotide repeat expansion disorders. It puts key points in the form of questions thus challenging the reader to retain knowledge. As board and certificate exams require knowledge of new technologies and applications this book helps users meet that challenge. Includes over 1 0000 multiple choice USMLE style questions to help readers prepare for specialty exams in Clinical Cytogenetics and Clinical Molecular Genetics. Designed to assist clinical molecular genetic fellows, genetic counselors, medical genetic residents and fellows, and molecular pathologist residents in preparing for their certification exam. Assists trainees on how to follow guidelines and put them in practice.

**Laboratory Experiments in General Chemistry**, 1974    Publications, Reports, and Papers for 1966 from Oak Ridge National Laboratory F. M. O'Hara, Ann S. Klein, 1967    **Science Educator's Guide to Laboratory Assessment** Rodney L. Doran, 2002 The book opens with an up to date discussion of assessment theory, research and uses. Then comes a wealth of sample assessment activities in biology, chemistry, physics and Earth science. Keyed to the National Science Education Standards, the activities include reproducible task sheets and scoring rubrics. All are ideal for helping students reflect on their own learning during science lab.

Foundations of Chemistry in the Laboratory Morris Hein, Leo R. Best, Robert L. Miner, 1973    Lab Experiments for Modern Chemistry Tzimopoulo, 1990    Science Directions 8 Douglas A. Roberts, Mary Kay Winter, 1990    **Medical Laboratory Science Review** Robert R. Harr, 2012-10-11 Use this comprehensive resource to gain the theoretical and practical knowledge you need to be prepared for classroom tests and certification and licensure examinations.    Practical Druggist and Pharmaceutical Review of Reviews Benjamin Lillard, 1908    Papers from the Geophysical Laboratory, Carnegie Institution of Washington, 1910    Laboratory and Classroom Exercises in General

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