

**Ionic Salts Simulation Lab****Learning Goals:**

- Understand how an ionic salt dissolves describing a saturated solution microscopically and macroscopically.
- Calculate solubility in grams/100ml
- Determine why some salts are more soluble than others
- Design and carry out a virtual lab to test a hypothesis

**Directions:** Go to

[http://phet.colorado.edu/simulations/sims.php?sim=Salts\\_and\\_Solubility](http://phet.colorado.edu/simulations/sims.php?sim=Salts_and_Solubility) and run the Salts and Solubility simulation. Take a few minutes to get use to how the simulation works. Then answer the questions below.

**Questions:**

- 1) Describe what you observe when you shake table salt into water. Test numbers of particles with different volumes.
  - a. When the salt dissolves, which element becomes the anion (the negatively charged atom) and which one becomes the cation (the positively charged atom)?
  - b. What is the relationship between amount of particles and amount of water? (direct or indirect).
  - c. At  $5.00 \times 10^{-23}$  what is the point of saturation (point where you can't dissolve any more salt)?
- 2) Fill in the chart below to see if the ratio of particles to water is constant at the saturation point.

Particles	Water Volume	Ratio (particles/volume)

- a. Calculate the constant below (show your work)

- 3) Test a variety of slightly soluble salts, noting the relationship between numbers of each ion pair. Fill in the chart below to help you see a relationship.
  - a. How many particles of each element need to be present before the salt stops dissolving (point of saturation)?

# Lab 30 Solubility Of A Salt Answers

**Frank Brescia**



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**Laboratory Guide to Pharmaceutical Technique** University of Wisconsin. Department of Pharmacy,1926

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Doran,2002 The book opens with an up to date discussion of assessment theory research and uses Then comes a wealth of sample assessment activities in biology chemistry physics and Earth science Keyed to the National Science Education Standards the activities include reproducible task sheets and scoring rubrics All are ideal for helping students reflect on their own learning during science lab Foundations of Chemistry in the Laboratory Morris Hein,Leo R. Best,Robert L.

Miner,1973 Lab Experiments for Modern Chemistry Tzimopoulo,1990 Science Directions 8 Douglas A. Roberts,Mary Kay Winter,1990 **Medical Laboratory Science Review** Robert R Harr,2012-10-11 Use this comprehensive resource to

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