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DETERMINING AN EMPIRICAL FORMULA

PROBLEM

Determining the elements that a mineral contains may be challenging, because some minerals contain the same elements, just in different proportions, as shown in the table below.

Some Silicon-Containing Minerals Found in Meteorites

| Mineral | Chemical formula |
|--------------|-------------------------------|
| Cristobalite | SiO_2 |
| Quartz | SiO_4 |
| Enstatite | MgSiO_3 |
| Forsterite | Mg_2SiO_4 |
| Diopside | $\text{MgCa}(\text{SiO}_3)_2$ |
| Monticellite | MgCaSiO_4 |

Background

Minerals such as the compounds in the table are actually formed in networks of repeating atoms, instead of as isolated molecules. To distinguish between two minerals with the same elements, such as cristobalite (SiO_2) and quartz (SiO_4), determine the ratios of the atoms. Cristobalite has two oxygen atoms for every silicon atom, and quartz has four oxygen atoms for every silicon atom.

Determining Empirical Formula Example

A mineral from a rock sample is analyzed. Its composition is 48.0% oxygen, 27.9% iron, and 24.1% sulfur. What is the empirical formula for this compound?

Analyze

Given: percentage composition: 48.0% O 27.9% Fe 24.1% S

Unknown: empirical formula

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N Colangelo



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