

MULTIPLE CHOICE. Choose the one alternative that best completes the statement or answers the question.

1)

Which of the following statements about the atom ${}^6\text{C}$ is FALSE?

- A) It has 12 neutrons in its nucleus.
- B) It has 6 protons in its nucleus.
- C) Its atomic weight is 12.
- D) It has 6 electrons orbiting the nucleus.
- E) Its atomic number is 6.

12)

13)

Table 2.1



2) Using the information in Table 2.1, calculate the molecular weight of ethanol, $\text{C}_2\text{H}_5\text{OH}$.

2)

- A) 33
- B) 96
- C) 46
- D) 34
- E) The answer cannot be determined.

3) Antacids neutralize acid by the following reaction. Identify the salt in the following equation:

3)



- A) H_2O
- B) MgCl_2
- C) HCl
- D) $\text{Mg}(\text{OH})_2$
- E) None of the answers is correct.

4) Which of the following statements is FALSE?

4)

- A) Water freezes from the top down.
- B) Salts readily dissolve in water.
- C) Water molecules are formed by hydrolysis.
- D) Water is a part of a dehydration synthesis reaction.
- E) Water is a polar molecule.

5) Which of the following is the type of bond holding K^+ and I^- ions in KI ?

5)

- A) hydrogen bond
- B) ionic bond
- C) covalent bond

6) Which of the following is the type of bond between molecules of water in a beaker of water?

6)

- A) hydrogen bond
- B) covalent bond
- C) ionic bond

7) What is the type of bond holding hydrogen and oxygen atoms in the H_2O molecule?

7)

- A) hydrogen bond
- B) ionic bond
- C) covalent bond

8) Identify the following reaction: $\text{Glucose} + \text{Fructose} \rightarrow \text{Sucrose} + \text{Water}$

8)

- A) reversible reaction
- B) exchange reaction
- C) dehydration synthesis reaction
- D) hydrolysis reaction

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William Kassler

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