

# **Study Guide –Chapter 1**

## **CHM1033**

### **Test # 1: (1.4 on Chapter 1, Chapters 2 and 3)**

1. Express the following numbers using scientific notation.
  - a) 351 000 000 000
  - b) 0.000 960
  - c) 5 207 000
  - d) 0.000 000 050
2. True or False?
  - a) The decimal number 0.0000010 expressed in scientific notation is  $1.0 \times 10^6$  \_\_\_\_\_
  - b) The decimal number 0.0210 expressed in scientific notation is  $2.10 \times 10^{-2}$  \_\_\_\_\_
  - c) The mass of an object,  $4.55 \times 10^3$  g, expressed in decimal notation is 0.00455 g \_\_\_\_\_
  - d) Light is an example of a chemical \_\_\_\_\_
3. a) A substance that consists of one type of matter and always has the same composition and properties is called a \_\_\_\_\_  
b) The study of substances that contain the element carbon is called \_\_\_\_\_  
c) Any material used in or produced by a chemical reaction is a \_\_\_\_\_
4. What is each number in scientific notation?
  - a) 0.00000001
  - b) 0.143
  - c) 12450000
  - d) 0.00546

# Miami Dade College Bat Study Guide

**Robert R. Redfield**

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