

## Heat Capacity Practice Problems-Worksheet 4

Name: \_\_\_\_\_ Period: \_\_\_\_\_

SHOW ALL WORK FOR FULL CREDIT! Remember to include units in your answers. Round answers to the proper number of significant figures.

Substance	Specific Heat (J/g x °C)
Water	4.18
Aluminum	0.90
Iron	0.46
Steam	1.7
Mercury	0.14
Chloroform	0.96
Grain Alcohol	2.4
Silver	0.24
Ice	2.1

1. How much heat will be absorbed by 25g of chloroform when its temperature started at 67.0°C and after it was 112.0°C?

Calculator answer: \_\_\_\_\_

SF answer: \_\_\_\_\_

2. How much heat will be absorbed by 130g of Mercury when its temperature started at 24.0°C and after it was 78.0°C?

Calculator answer: \_\_\_\_\_

SF answer: \_\_\_\_\_

3. If the temperature of a 95.00 g sample of Iron started at 36.0°C and ended at 58.0°C, how much heat would be lost?

Calculator answer: \_\_\_\_\_

SF answer: \_\_\_\_\_

4. What is the definition of Temperature? \_\_\_\_\_  
\_\_\_\_\_

5. What is the definition of Heat? \_\_\_\_\_  
\_\_\_\_\_

6. Why can there never be a temperature less than 0 Kelvins? \_\_\_\_\_  
\_\_\_\_\_

# Heat Capacity Practice Problems 4 Answer Key

**Albert A Gayle**

## **Heat Capacity Practice Problems 4 Answer Key:**

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**Measurements** Richard A. Robie, 1972 Additional title page description A description of the construction and operation of a calorimeter for determining the heats of solution in aqueous HF or HCl and of a low temperature heat capacity calorimeter for Cp measurements between 10 and 380 kelvin     *Comprehensive Chemistry XI* Dr. B. Kapila, S. K. Khanna, 2010-11

Comprehensive chemistry according to the new syllabus prescribed by Central Board of Secondary Education CBSE     **A**

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