

Heat Capacity Practice Problems-Worksheet 4

Name: _____ Period: _____

SHOW ALL WORK FOR FULL CREDIT! Remember to include units in your answers. Round answers to the proper number of significant figures.

Substance	Specific Heat (J/g x °C)
Water	4.18
Aluminum	0.90
Iron	0.46
Steam	1.7
Mercury	0.14
Chloroform	0.96
Grain Alcohol	2.4
Silver	0.24
Ice	2.1

1. How much heat will be absorbed by 25g of chloroform when its temperature started at 67.0°C and after it was 112.0°C?

Calculator answer: _____

SF answer: _____

2. How much heat will be absorbed by 130g of Mercury when its temperature started at 24.0°C and after it was 78.0°C?

Calculator answer: _____

SF answer: _____

3. If the temperature of a 95.00 g sample of Iron started at 36.0°C and ended at 58.0°C, how much heat would be lost?

Calculator answer: _____

SF answer: _____

4. What is the definition of Temperature? _____

5. What is the definition of Heat? _____

6. Why can there never be a temperature less than 0 Kelvins? _____

Heat Capacity Practice Problems 4 Answer Key

Albert A Gayle

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Heat Capacity Practice Problems 4 Answer Key:

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