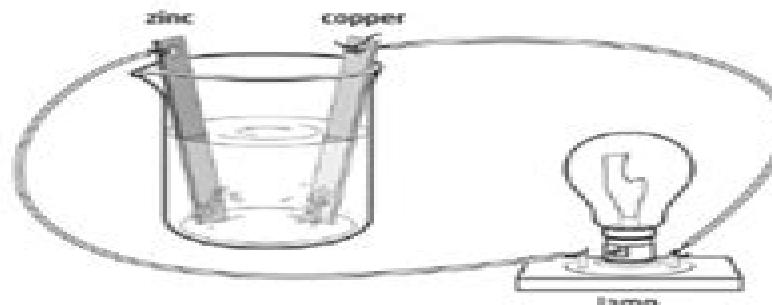


# Chapter 11 Workbook

## Section 11.1 Review - Cells and Batteries

### Multiple Choice

- Which two materials would make good electrodes for a voltaic (wet) cell?
  - copper and zinc
  - two strips of copper metal in distilled water
  - acid and copper
  - acid and zinc
  - two strips of zinc
- Which of the following is *not* an advantage of using a dry cell over a wet cell?
  - Dry cells are less easily damaged compared with wet cells.
  - Dry cells are easier to handle than voltaic cells.
  - Dry cells are used in portable radios.
  - Dry cells can be connected to make a battery, but wet cells cannot make a battery.
  - Dry cells can be transported more easily.
- What is a hybrid vehicle?
  - a vehicle that is fuelled by solar energy
  - a vehicle that is fuelled by solar energy and hydroelectric energy
  - a vehicle that is fuelled by two different types of fossil fuels
  - a vehicle that is fuelled by fossil fuels and electrical energy
  - a vehicle that is fuelled by propane only
- What types of cells can power a flashlight?
  - zinc-carbon cells
  - alkaline cells
  - rechargeable cells
  - only a voltaic wet cell
  - any of A, B, and C
- Which of the following is *true* for a voltaic cell?
  - The electrodes must be made of metal.
  - The electrolyte must be conducting.
  - The electrodes must be an insulator.
  - A and B are true.
  - B and C are true.
- Which cell has the least impact on the environment?
  - wet cell
  - dry cell
  - solar cell
  - fuel cell
  - alkaline cell
- What type of cell is a lead-acid cell?
  - primary dry cell
  - primary wet cell
  - secondary dry cell
  - secondary wet cell
  - none of the above
- Which of the following statements is *true* about the diagram below?



- The zinc electrode forms positive ions, and electrons flow from zinc to copper through the lamp.
- The zinc electrode forms negative ions, and electrons flow from zinc to copper through the lamp.
- The zinc electrode forms positive ions, and electrons flow from copper to zinc through the lamp.
- The zinc electrode forms positive ions, and protons flow from copper to zinc through the lamp.
- The zinc electrode forms negative ions, and electrons flow from copper to zinc through the lamp.

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**Gary W. Heiman**

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1. Understanding the eBook Interactive Science Workbook Ch11
  - The Rise of Digital Reading Interactive Science Workbook Ch11
  - Advantages of eBooks Over Traditional Books
2. Identifying Interactive Science Workbook Ch11
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Interactive Science Workbook Ch11
  - User-Friendly Interface
4. Exploring eBook Recommendations from Interactive Science Workbook Ch11
  - Personalized Recommendations
  - Interactive Science Workbook Ch11 User Reviews and Ratings
  - Interactive Science Workbook Ch11 and Bestseller Lists
5. Accessing Interactive Science Workbook Ch11 Free and Paid eBooks
  - Interactive Science Workbook Ch11 Public Domain eBooks
  - Interactive Science Workbook Ch11 eBook Subscription Services

- Interactive Science Workbook Ch11 Budget-Friendly Options
- 6. Navigating Interactive Science Workbook Ch11 eBook Formats
  - ePUB, PDF, MOBI, and More
  - Interactive Science Workbook Ch11 Compatibility with Devices
  - Interactive Science Workbook Ch11 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Interactive Science Workbook Ch11
  - Highlighting and Note-Taking Interactive Science Workbook Ch11
  - Interactive Elements Interactive Science Workbook Ch11
- 8. Staying Engaged with Interactive Science Workbook Ch11
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Interactive Science Workbook Ch11
- 9. Balancing eBooks and Physical Books Interactive Science Workbook Ch11
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Interactive Science Workbook Ch11
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Interactive Science Workbook Ch11
  - Setting Reading Goals Interactive Science Workbook Ch11
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Interactive Science Workbook Ch11
  - Fact-Checking eBook Content of Interactive Science Workbook Ch11
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends

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