

# Half Modern Chemistry Review

## CHAPTER 5: THE PERIODIC TABLE

The following pages contain the **best** (but not all) of the information for this chapter is that.

Focus on this content, but make sure to review other topics: reactions, bonding, molecules, etc.

If you study this document and **THOROUGHLY** study, you should at least be able to PASS this test.

Good luck with the exam, exercises, and/or application of this content. ~~Good~~

### OUTLINE:

- Cells/molecules with periodic table characteristics/properties: **PC 12**
- Cells/molecules with periodic table characteristics/properties: **PC 13**
- 5.1: Elements that represent all the modern Periodic table: **PC 14** (periodic table, and groups: **PT 11** & **12**)
- 5.2: How the Periodic Table is related to chemical properties: **PC 15** (periodic table, and chemical properties, and chemical properties, and chemical properties: **PT 11** & **12**)
- 5.3: Identify elements: **PC 16** (periodic table, and chemical properties: **PT 11** & **12**)
- 5.4: Compare periodic table: **PC 17** (periodic table, and chemical properties: **PT 11** & **12**)

### 5.1: HISTORY OF THE PERIODIC TABLE

- **Introduction**
  - **periodic table** = the table that shows that the repeating chemical and physical properties of elements change periodically with the atomic number of the element.
  - **periodic table** = an arrangement of the elements in order of their atomic numbers such that elements with similar properties fall in the same column.
  - **periodic table** = a number of the table with numbers of elements, chemical properties, and other data.
  - **periodic table** = a table of the elements of the periodic table that shows that elements with similar properties fall in the same column.
- **Chapter 5.1.1: Introduction**
  - The periodic table shows that the physical and chemical properties of the elements, and periodic properties, of the elements, change.
  - The periodic table is an arrangement of the elements in order of their atomic numbers such that elements with similar properties fall in the same column/group/family.
  - The elements in the periodic table are ordered by atomic number.

### 5.2: ELECTRON CONFIGURATIONS AND THE PERIODIC TABLE

- **Introduction**
  - **atomic number** = one of the elements of Group 1 of the periodic table (hydrogen, sodium, potassium, rubidium, cesium, and francium).
  - **atomic number** = one of the elements of Group 2 of the periodic table (beryllium, magnesium, calcium, strontium, barium, and radium).
  - **atomic number** = one of the elements of Group 13 of the periodic table (boron, aluminum, gallium, indium, and thallium).
  - **atomic number** = one of the elements of Group 14 of the periodic table (carbon, silicon, germanium, tin, and lead).
  - **atomic number** = one of the elements of Group 15 of the periodic table (nitrogen, phosphorus, arsenic, antimony, and bismuth).

# Holt Modern Chemistry Answer Key Review

**J Ma**



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