

Assessment**Chapter Test A****Chapter: Acids and Bases**

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

_____ 1. Which of the properties listed below is *not* characteristic of an acid?
a. a sour taste
b. a slippery feel
c. the ability to conduct an electric current
d. reactivity with metals

_____ 2. Which is an example of a weak acid?
a. HCl
b. H₂SO₄
c. H₃PO₄
d. HNO₃

_____ 3. Acetic acid is found in
a. vinegar.
b. the stomach.
c. antacids.
d. oranges.

_____ 4. A characteristic of an Arrhenius base is that it
a. is an electrolyte.
b. tastes bitter.
c. dissociates to form OH⁻ ions.
d. All of the above

_____ 5. Which of the following is a Brønsted-Lowry base?
a. HCl
b. HCO₃⁻
c. H₃O⁺
d. H₃PO₄

_____ 6. In the reaction represented by the equation
$$\text{H}_2\text{C}_2\text{O}_4(\text{aq}) + \text{CH}_3\text{NH}_2(\text{aq}) \rightleftharpoons \text{HC}_2\text{O}_4(\text{aq}) + \text{CH}_3\text{NH}_3^+(\text{aq}),$$
which of these is a conjugate acid-base pair?
a. H₂C₂O₄(aq) and CH₃NH₂(aq)
b. H₂C₂O₄(aq) and CH₃NH₃⁺(aq)
c. CH₃NH₂(aq) and CH₃NH₃⁺(aq)
d. HC₂O₄(aq) and CH₃NH₃⁺(aq)

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Wade A. Freeman

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