

Electron Configuration Multiple Choice Questions for Unit Review

Multiple Choice Questions

For each of the following questions, choose the letter that **best** answers the question.

[Show questions one by one](#)

1. How many electrons are there in the **third** shell (principal energy level) of the atom with atomic number 23?

- A. 10
- B. 8
- C. 3
- D. 13
- E. 11

2. How many electrons are there in the "p" orbitals of the **last** shell (principal energy level) of silicon, atomic number 14?

- A. 1
- B. 0
- C. 3
- D. 2
- E. 4

3. Which of the following is the correct order for electrons filling in orbitals?

- A. 2p, 3s, 3p, 3d
- B. 4s, 3d, 4p, 4d
- C. 3s, 3p, 4s, 3d
- D. 1s, 2s, 2p, 2d
- E. 3p, 4s, 3d, 5s

Electron Configuration Multiple Choice Questions

**Dr Vijay Sarda, Dr A C Handa, Dr K K
Arora**

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