



HSR

Tuning Manual

Revised 02/10/03

Mikuni Tuning Manual

Olav Aaen



Mikuni Tuning Manual:

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web log $h = 10^{-4.05891} \times 10^{-5} \text{ mol dm}^{-3}$ log pK_a $pH = 3.75405$ 0.3010030050 and $hCOOH = 0.1$ so $hCOO^- = 0.200 \text{ mol dm}^{-3}$ amount vol x

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web only the better candidates could write successive ionisation equations in part c while part d about the pH of equimolar strong and weak acid solutions was poorly answered

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