

MCQ

Select the correct answer for the following multiple choice type questions.

1. The molar mass of oxalic acid is
 - a. 126 g
 - b. 12.6 g
 - c. 126 u
 - d. 126 g/mol
2. To prepare 1 M solution of oxalic acid
 - a. 126 g of oxalic acid dissolve in distilled water and diluted to 1 L
 - b. 12.6 g of oxalic acid dissolve in distilled water and diluted to 1 L
 - c. 1.26 g of oxalic acid dissolve in distilled water and diluted to 1 L
 - d. 0.126 g of oxalic acid dissolve in distilled water and diluted to 1 L
3. Water of crystallization present in oxalic acid is/are
 - a. 1
 - b. 2
 - c. 3
 - d. 5
4. Oxalic acid is used to prepare standard solution because it is a
 - a. substance
 - b. primary standard substance
 - c. secondary standard substance
 - d. tertiary standard substance
5. The quantity of oxalic acid required to prepare 0.1 M 100 mL standard solution of oxalic acid is
 - a. 126 g
 - b. 12.6 g
 - c. 1.26 g
 - d. 0.126 g

Short answer questions

1. Calculate the molar mass of oxalic acid. (H = 1, C = 12, O = 16)
 Ans. Molar mass of $\text{H}_2\text{C}_2\text{O}_4 = (2 \times 1) + (2 \times 12) + (4 \times 16)$
 $= 2 + 24 + 64$
 $= 90 \text{ g/mol}$
2. Why heating is required in oxalic acid and potassium permanganate titration?
 Ans. The solution is heated before titration because this reaction happens only at certain temp so oxalic acid is heated to speed up the liberation of Mn^{2+}
3. Why one test tube of 2M sulphuric acid is required in permanganate titration?
 Ans. To prevent oxidation of manganese to form manganese dioxide
4. What is the oxidation state of carbon atom in oxalic acid after completion of redox titration?
 Ans. The oxidation state of carbon atom in oxalic acid changes from +3 to +4
5. Why primary standard substances are used to make standard solutions?
 Ans. Primary standard substance are used in titration to determine unknown concentration

Remark and signs of teacher:

Give Me D Answer Of Chemistry Pratical Alternative A

L Darling-Hammond

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