

MCQ

Select the correct answer for the following multiple choice type questions.

- The molar mass of oxalic acid is
 a. 126 u ☐ ☒ 126 g/mol
 c. 12.6 u ☐ d. 12.6 g/mol ☐
- To prepare 1 M solution of oxalic acid
 a. 126 g of oxalic acid dissolve in distilled water and diluted to 1 L
 b. 12.6 g of oxalic acid dissolve in distilled water and diluted to 1 L
 c. 1.26 g of oxalic acid dissolve in distilled water and diluted to 1 L
 d. 0.126 g of oxalic acid dissolve in distilled water and diluted to 1 L
- Water of crystallization present in oxalic acid is/are.....
 a. 1 ☒ 2 ☐ 3 ☐ 5 ☐
- Oxalic acid is used to prepare standard solution because it is a.....
 a. substance ☐ b. primary standard substance ☐
 c. secondary standard substance ☐ d. tertiary standard substance ☐
- The quantity of oxalic acid required to prepare 0.1 M 100 mL standard solution of oxalic acid is
 a. 126 g ☐ b. 12.6 g ☐ ☒ 1.26 ☐ d. 0.126 g ☐

Short answer questions

1. Calculate the molar mass of oxalic acid. (H = 1, C = 12, O = 16)

Ans Molar mass of $H_2C_2O_4 = (2 \times 1) + (2 \times 12) + (4 \times 16)$
 $= 2 + 24 + 64$
 $= 90 \text{ g/mol}$

2. Why heating is required in oxalic acid and potassium permanganate titration?

Ans The solution is heated before titration because this reaction happens only at certain temp so oxalic acid is heated to speed up the liberation of Mn^{2+}

3. Why one test tube of 2M sulphuric acid is required in permanganate titration?

Ans To prevent oxidation of manganese to form manganese dioxide

4. What is the oxidation state of carbon atom in oxalic acid after completion of redox titration?

Ans The oxidation state of carbon atom in oxalic acid change from +3 to +4

5. Why primary standard substances are used to make standard solutions?

Ans Primary standard substance are used in titration to determine unknown concentration

Remark and sign of teacher: _____

Give Me D Answer Of Chemistry Pratical Alternative A

L Darling-Hammond



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