
Activity A: Heat transfer	Get the Gizmo ready: ● Click Reset (↻).	
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Question: What factors determine how heat energy transfers between objects?

1. **Predict:** In the Gizmo warm-up, you saw how 200 g of 90 °C copper transfers heat to 200 g of 30.0 °C water.

A. How do you think increasing the water's mass would affect the final temperature?

I think that will make the temperature drop

B. How do you think decreasing the copper's mass would affect the final temperature?

I think temperature will also drop

C. How do you think increasing or decreasing the copper's initial temperature would affect the final temperature? It wouldn't affect it that much

since the specific heat capacity is larger on water

2. **Collect data:** Use the Gizmo to determine the final temperature for each set-up listed below. Record your results in the tables. In the first table, you experiment with changing the water's mass. In the second table, you change the copper's mass. In the third table, you change the initial temperature of the copper. The first row of each table has been completed for you.

Copper		Water		Final Temp. (°C)
Initial Temp. (°C)	Mass (g)	Initial Temp. (°C)	Mass (g)	
90 °C	200 g	30.0 °C	200 g	34.96 °C
90 °C	200 g	30.0 °C	2,000 g	30.54
90 °C	200 g	30.0 °C	200 g	34.96 °C
90 °C	20 g	30.0 °C	200 g	30.54



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Table of Contents Gizmo Answer Key To Calorimetry

1. Understanding the eBook Gizmo Answer Key To Calorimetry
 - The Rise of Digital Reading Gizmo Answer Key To Calorimetry
 - Advantages of eBooks Over Traditional Books
2. Identifying Gizmo Answer Key To Calorimetry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gizmo Answer Key To Calorimetry
 - User-Friendly Interface
4. Exploring eBook Recommendations from Gizmo Answer Key To Calorimetry
 - Personalized Recommendations
 - Gizmo Answer Key To Calorimetry User Reviews and Ratings
 - Gizmo Answer Key To Calorimetry and Bestseller Lists
5. Accessing Gizmo Answer Key To Calorimetry Free and Paid eBooks

- Gizmo Answer Key To Calorimetry Public Domain eBooks
- Gizmo Answer Key To Calorimetry eBook Subscription Services
- Gizmo Answer Key To Calorimetry Budget-Friendly Options

6. Navigating Gizmo Answer Key To Calorimetry eBook Formats

- ePUB, PDF, MOBI, and More
- Gizmo Answer Key To Calorimetry Compatibility with Devices
- Gizmo Answer Key To Calorimetry Enhanced eBook Features

7. Enhancing Your Reading Experience

- Adjustable Fonts and Text Sizes of Gizmo Answer Key To Calorimetry
- Highlighting and Note-Taking Gizmo Answer Key To Calorimetry
- Interactive Elements Gizmo Answer Key To Calorimetry

8. Staying Engaged with Gizmo Answer Key To Calorimetry

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Gizmo Answer Key To Calorimetry

9. Balancing eBooks and Physical Books Gizmo Answer Key To Calorimetry

- Benefits of a Digital Library
- Creating a Diverse Reading Collection Gizmo Answer Key To Calorimetry

10. Overcoming Reading Challenges

- Dealing with Digital Eye Strain
- Minimizing Distractions
- Managing Screen Time

11. Cultivating a Reading Routine Gizmo Answer Key To Calorimetry

- Setting Reading Goals Gizmo Answer Key To Calorimetry
- Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Gizmo Answer Key To Calorimetry

- Fact-Checking eBook Content of Gizmo Answer Key To Calorimetry
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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