

## Chapter 5 Assessment

pages 166–169

### Section 5.1

#### Mastering Concepts

34. Define the following terms.

a. frequency

Frequency is the number of waves that pass a given point per second.

b. wavelength

Wavelength is the shortest distance between equivalent points on a continuous wave.

c. quantum

A quantum is the minimum amount of energy that can be lost or gained by an atom.

d. ground state

An atom's ground state is its lowest allowable energy state.

35. Arrange the following types of electromagnetic radiation in order of increasing wavelength.

a. ultraviolet light

b. microwaves

c. radio waves

d. X rays

d. X rays, a. ultraviolet light, b. microwaves, c. radio waves

36. A gamma ray has a frequency of  $2.88 \times 10^{21}$  Hz. What does that mean?

$2.88 \times 10^{21}$  gamma-ray electromagnetic waves of this frequency pass a given point per second.

37. What is the photoelectric effect?

a phenomenon in which a metal emits electrons when light of a sufficient frequency shines on it

38. **Neon Sign** How does light emitted from a neon sign differ from sunlight?

The light from a neon sign contains only certain visible colors, while sunlight contains the full spectrum of colors.

39. Explain Planck's quantum concept as it relates to energy lost or gained by matter.

According to Planck, for a given frequency,  $\nu$ , matter can emit or absorb energy only in discrete quanta that are whole-number multiples of  $\hbar\nu$ , where  $\hbar$  is Planck's constant.

40. How did Einstein explain the photoelectric effect?

He proposed that photons must have a certain minimum energy level, or threshold, value to cause the ejection of a photoelectron.

41. **Rainbow** What are two differences between the red and green electromagnetic waves in a rainbow?

The red waves have a longer wavelength and a lower frequency.

42. **Temperature** What happens to the light emitted by a heated, glowing object as its temperature increases?

The color of the light changes as the object acquires more and more energy.

43. What are three deficiencies of the wave model of light related to light's interaction with matter?

The wave model does not explain the photoelectric effect, atomic emission spectra, and why matter emits different frequencies of light at different temperatures.

44. How are radio waves and ultraviolet waves similar? How are they different?

Both types of waves travel at the same speed in a vacuum,  $3.00 \times 10^8$  m/s. Radio waves have a longer wavelength and lower frequency than ultraviolet waves.

# Glencoe Chemistry Matter Change Answer Key Chapter 5

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**Chemistry** McGraw-Hill Staff,2001-03 **Technical questions and answers for job interview Offshore Oil & Gas Platforms** Petrogav International Oil & Gas Training

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Glencoe Earth Science Ralph M. Feather,1999 Earth science is the study of Earth and space It is the study of such things

as the transfer of energy in Earth's atmosphere the evolution of landforms patterns of change that cause weather the scale and structure of stars and the interactions that occur among the water atmosphere and land Earth science in this book is divided into four specific areas of study geology meteorology astronomy and oceanography p 8 9

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Staff,Glencoe/McGraw-Hill,2001-04-01    **Properties of Matter: Physical Changes vs. Chemical Changes** Gr. 5-8 George Graybill,2015-09-01 This is the chapter slice Physical Changes vs Chemical Changes from the full lesson plan Properties of Matter Discover what matter is and is not Learn about and the difference between a mixture and a solution Chocked full with hands on activities to understand the various physical and chemical changes to matter Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Written to grade these science concepts are presented in a way that makes them more accessible to students and easier to understand Our resource is jam packed with experiments reading passages and activities all for students in grades 5 to 8 Color mini posters and answer key included and can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom's Taxonomy and STEM initiatives

**Properties of Matter: Mixtures and Solutions** Gr. 5-8

George Graybill,2015-09-01 This is the chapter slice Mixtures and Solutions from the full lesson plan Properties of Matter Discover what matter is and is not Learn about and the difference between a mixture and a solution Chocked full with hands on activities to understand the various physical and chemical changes to matter Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Written to grade these science concepts are presented in a way that makes them more accessible to students and easier to understand Our resource is jam packed with experiments reading passages and activities all for students in grades 5 to 8 Color mini posters and answer key included and can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom's Taxonomy and STEM initiatives

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*Supplemental Problems - Glencoe Chemistry Matter and Change* ,2013

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