

Unit 7: Distinguishing Chemical Reactions - Worksheets 2

Directions: Below are 10 reactions. For each reaction, determine the type of reaction.

1. $\text{H}_2\text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_2$ _____
2. $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$ _____
3. $\text{H}_2\text{O} + \text{CO}_2 \rightarrow \text{H}_2\text{CO}_3$ _____
4. $\text{H}_2\text{O}_2 \rightarrow \text{H}_2\text{O}$ _____
5. $\text{H}_2\text{O} + \text{CO}_2 \rightarrow \text{H}_2\text{CO}_3$ _____
6. $\text{H}_2\text{O} + \text{CO}_2 \rightarrow \text{H}_2\text{O}$ _____
7. $\text{H}_2\text{O}_2 \rightarrow \text{H}_2\text{O}$ _____
8. $\text{H}_2\text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_2$ _____
9. $\text{H}_2\text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_2$ _____
10. $\text{H}_2\text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_2$ _____

These reactions must be balanced and written in a balanced form. Then, determine the type.

11. $\text{H}_2\text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_2$ _____
12. $\text{CH}_3\text{CH}_2\text{OH} \rightarrow \text{CH}_3\text{CH}_2\text{O} + \text{H}_2\text{O}$ _____

13. $\text{CH}_3\text{CH}_2\text{OH} + \text{C}_2\text{H}_5\text{OH} \rightarrow \text{CH}_3\text{CH}_2\text{OCH}_2\text{CH}_3 + \text{H}_2\text{O}$ _____
14. $\text{CH}_3\text{CH}_2\text{OH} \rightarrow \text{CH}_3\text{CH}_2\text{O} + \text{H}_2\text{O}$ _____
15. $\text{CH}_3\text{CH}_2\text{OH} \rightarrow \text{CH}_3\text{CH}_2\text{O} + \text{H}_2\text{O}$ _____
16. $\text{CH}_3\text{CH}_2\text{OH} \rightarrow \text{CH}_3\text{CH}_2\text{O} + \text{H}_2\text{O}$ _____
17. $\text{CH}_3\text{CH}_2\text{OH} \rightarrow \text{CH}_3\text{CH}_2\text{O} + \text{H}_2\text{O}$ _____
18. $\text{CH}_3\text{CH}_2\text{OH} \rightarrow \text{CH}_3\text{CH}_2\text{O} + \text{H}_2\text{O}$ _____

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G Thomas

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