

## CK-12 Chemistry - Basic Answer Key

### Chapter 7: Chemical Nomenclature

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#### 7.1 Ionic Compounds

##### Lesson Review Questions

##### Questions

- What is the purpose of an empirical formula?
- Define binary and ternary ionic compounds.
- What is the difference between a monatomic and a polyatomic ion?
- For each of the following ionic compounds, what ions are present and in what ratio?
  - $\text{MgBr}_2$
  - $\text{Li}_2\text{CO}_3$
  - $\text{Fe}_2(\text{SO}_4)_3$
- Predict the compound that forms when gallium combines with oxygen. What would the name of this compound be?
- Give the formula for each of the following ionic compounds:
 

(a) ammonium nitrate	(d) vanadium(III) oxide
(b) cobalt(II) sulfate	(e) barium oxide
(c) nickel(II) cyanide	(f) calcium hypochlorite
- Name the following ionic compounds:
  - $\text{MgBr}_2$
  - (d)  $\text{KMnO}_4$
  - $\text{Li}_2\text{CO}_3$
  - $(\text{NH}_4)_2\text{S}$
  - $\text{KHSO}_3$
  - $\text{CuCl}$
  - $\text{CuCl}_2$
- Write the correct formulas for the following ionic compounds:
  - Barium chloride
  - Chromium(III) oxide
  - Potassium sulfate
  - Zinc phosphate

##### Answer:

- The empirical formula is the most straightforward way to describe the structure of a molecule, using a chemical formula that gives the lowest whole number ratio between the ions involved.

# Mastering Chemistry Answer Key Chapter 7

**Kaplan, Kaplan Educational Center  
Staff**



## **Mastering Chemistry Answer Key Chapter 7:**

**A Problem-Solving Approach to Aquatic Chemistry** James N. Jensen, 2023-01-05 A Problem Solving Approach to Aquatic Chemistry Enables civil and environmental engineers to understand the theory and application of aquatic equilibrium chemistry The second edition of A Problem Solving Approach to Aquatic Chemistry provides a detailed introduction to aquatic equilibrium chemistry calculation methods for systems at equilibrium applications of aquatic chemistry and chemical kinetics The text directly addresses two required ABET program outcomes in environmental engineering chemistry including stoichiometry equilibrium and kinetics and material and energy balances fate and transport of substances in and between air water and soil phases The book is very student centered with each chapter beginning with an introduction and ending with a summary that reviews the chapter's main points To aid in reader comprehension important terms are defined in context and key ideas are summarized Many thought provoking discussion questions worked examples and end of chapter problems are also included Each part of the text begins with a case study a portion of which is addressed in each subsequent chapter illustrating the principles of that chapter In addition each chapter has an Historical Note exploring connections with the people and cultures connected to topics in the text A Problem Solving Approach to Aquatic Chemistry includes Fundamental concepts such as concentration units thermodynamic basis of equilibrium and manipulating equilibria Solutions of chemical equilibrium problems including setting up the problems and algebraic graphical and computer solution techniques Acid base equilibria including the concepts of acids and bases titrations and alkalinity and acidity Complexation including metals ligands equilibrium calculations with complexes and applications of complexation chemistry Oxidation reduction equilibria including equilibrium calculations graphical approaches and applications Gas liquid and solid liquid equilibrium with expanded coverage of the effects of global climate change Other topics including chemical kinetics of aquatic systems surface chemistry and integrative case studies For advanced senior undergraduates and first year graduate students in environmental engineering courses A Problem Solving Approach to Aquatic Chemistry serves as an invaluable learning resource on the topic with a variety of helpful learning elements included throughout to ensure information retention and the ability to apply covered concepts in practical settings

**Study Guide for The Human Body in Health and Illness**

Barbara Herlihy, 2013-11-27 Corresponding to the chapters in The Human Body in Health and Illness 4th Edition by Barbara Herlihy this study guide offers fun and practical exercises to help you review understand and remember basic A P Even if you find science intimidating this book can help you succeed Each chapter includes three parts Mastering the Basics with matching ordering labeling diagram reading and coloring exercises Putting It All Together including multiple choice quizzes and case studies Challenge Yourself with critical thinking questions and puzzles Textbook page references are included with the questions to make it easier to review difficult topics Objectives at the beginning of each chapter reinforce the goals of the textbook and set a framework for study UPDATED content matches the new and revised material in the 5th edition of the

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**Chemistry** James C. Hill,2003 This book assists students through the text material with chapter overviews learning objectives review of key terms cumulative chapter review quizzes and self tests Included are answers to all Student Guide exercises Chapter summaries are correlated to those in the Instructor s Resource Manual

Student's Guide to Brown and LeMay Chemistry James C. Hill,1988

**New York State Regents Exam** Kaplan,Kaplan Educational Center Staff,2000-02 Kaplan s guides to the New York State Regents Exams come complete with a comprehensive review of all the tested material plus Kaplan s exclusive test taking strategies This powerful combination makes the New York State Regents Exam Chemistry Second Edition a highly effective way for you to score higher on this very challenging test Are you ready for the New York State Regents Chemistry exam You will be with Kaplan s proven plan for success STEP 1 Take a Diagnostic Test The results of this test will outline your strengths and weaknesses You will find out the exact areas on which you need to focus your preparation STEP 2 Review the Tested Material Kaplan takes you through each section step by step providing you with effective tips and strategies to successfully answer every type of question STEP 3 Practice with Real Regents Exams Practice makes perfect And with this book you ll practice with the real thing actual Regents exams including detailed explanations for every answer and an analysis of your performance STEP 4 Succeed on the Test Follow Kaplan s plan for success on the Regents and you will score higher In fact we guarantee it See details inside

**Chemistry - The Central Science** James C. Hill,Bruce Edward Bursten,2006 Chemistry The Central Science is the most trusted book on the market its scientific accuracy clarity innovative pedagogy functional problem solving and visuals set this book apart Brown LeMay and Bursten teach students the concepts and skills they need without overcomplicating the subject A comprehensive media package that works in tandem with the text helps students practice and learn while providing instructors the tools they need to succeed Publisher s description

Find Your Ideal Match Khushabu Gupta,2025-12-10 Unlock the secrets to lasting love with Find Your Ideal Match The 10 Day Guide to Picking the Right Guy and Building Lasting Love by Khushabu Gupta This transformative guide is designed for women seeking meaningful relationships offering a step by step 10 day plan to help you identify attract and connect with the person who truly

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### **Merrill Chemistry**

Robert C. Smoot,Smoot,Richard G. Smith,Jack Price,1998

### **Modern**

*University Chemistry, Custom Pub* Norbert T. Porile,1993

### **Introductory Chemistry**

Charles H. Corwin,2005 For one semester courses in Basic Chemistry Introduction to Chemistry and Preparatory Chemistry and the first term of Allied Health Chemistry This text is carefully crafted to help students learn chemical skills and concepts more effectively Corwin covers math and problem solving early in the text he builds student confidence and skills through innovative problem solving pedagogy and technology formulated to meet student needs

### **An Introduction to the World's Oceans**

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**Instructor's Manual and Test Bank to Accompany Basic Concepts of Chemistry** Malone,2000-04-18

Books in Print Supplement ,2002

## Mastering Chemistry Answer Key Chapter 7 Book Review: Unveiling the Power of Words

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