

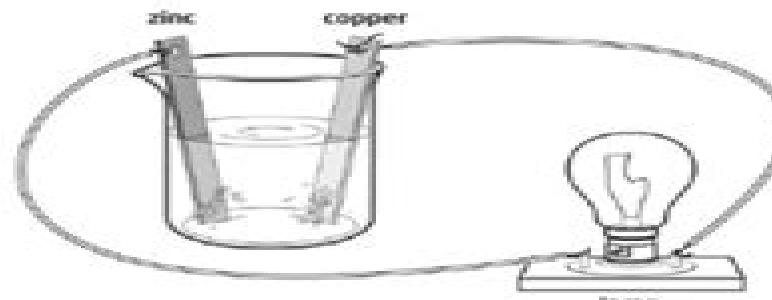
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# Chapter 11 Workbook

## Section 11.1 Review - Cells and Batteries

### Multiple Choice

- Which two materials would make good electrodes for a voltaic (wet) cell?
  - copper and zinc
  - two strips of copper metal in distilled water
  - acid and copper
  - acid and zinc
  - two strips of zinc
- Which of the following is *not* an advantage of using a dry cell over a wet cell?
  - Dry cells are less easily damaged compared with wet cells.
  - Dry cells are easier to handle than voltaic cells.
  - Dry cells are used in portable radios.
  - Dry cells can be connected to make a battery, but wet cells cannot make a battery.
  - Dry cells can be transported more easily.
- What is a hybrid vehicle?
  - a vehicle that is fuelled by solar energy
  - a vehicle that is fuelled by solar energy and hydroelectric energy
  - a vehicle that is fuelled by two different types of fossil fuels
  - a vehicle that is fuelled by fossil fuels and electrical energy
  - a vehicle that is fuelled by propane only
- What types of cells can power a flashlight?
  - zinc-carbon cells
  - alkaline cells
  - rechargeable cells
  - only a voltaic wet cell
  - any of A, B, and C
- Which of the following is *true* for a voltaic cell?
  - The electrodes must be made of metal.
  - The electrolyte must be conducting.
  - The electrodes must be an insulator.
  - A and B are true.
  - B and C are true.
- Which cell has the least impact on the environment?
  - wet cell
  - dry cell
  - solar cell
  - fuel cell
  - alkaline cell
- What type of cell is a lead-acid cell?
  - primary dry cell
  - primary wet cell
  - secondary dry cell
  - secondary wet cell
  - none of the above
- Which of the following statements is *true* about the diagram below?



- The zinc electrode forms positive ions, and electrons flow from zinc to copper through the lamp.
- The zinc electrode forms negative ions, and electrons flow from zinc to copper through the lamp.
- The zinc electrode forms positive ions, and electrons flow from copper to zinc through the lamp.
- The zinc electrode forms positive ions, and protons flow from copper to zinc through the lamp.
- The zinc electrode forms negative ions, and electrons flow from copper to zinc through the lamp.

# Interactive Science Workbook Ch11

**Ruth Ravid,Elizabeth Oyer**

## Interactive Science Workbook Ch11:

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