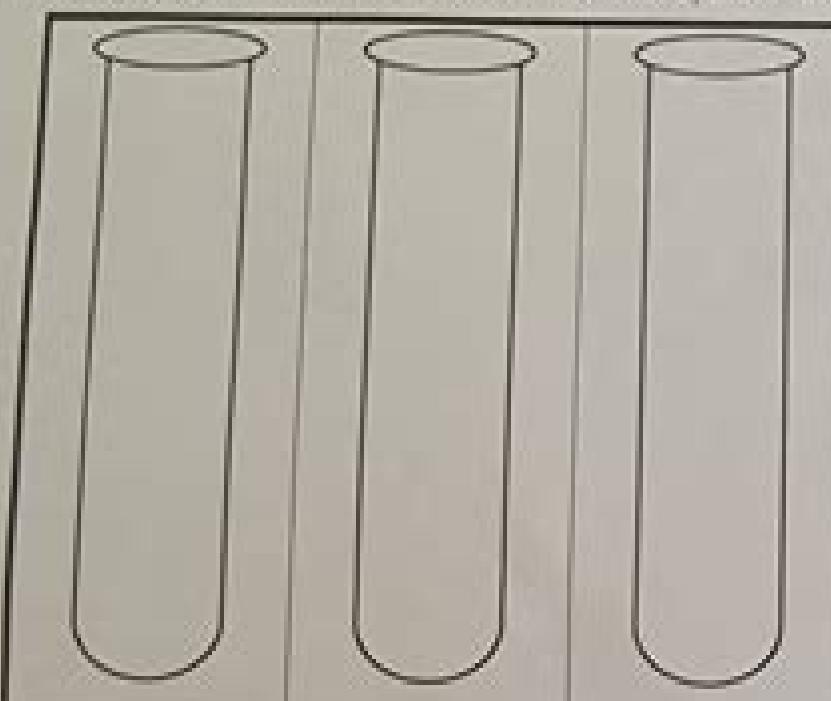


# Chemistry 162 Nine Solutions Problem – Post Lab (Upload as an image to Canvas)

For each set of reactants shown: 1) Create small particle drawings of the separate reactant solutions and the solution after mixing. Maintain stoichiometry and include at least 2  $\text{H}_2\text{O}$  molecules for (aq) mixtures 2) Write the balanced molecular reaction. 3) Write the balanced complete ionic equation (no cross-outs). 4) Write the balanced net ionic equation. Include phases for all species. See the lab manual for an example.

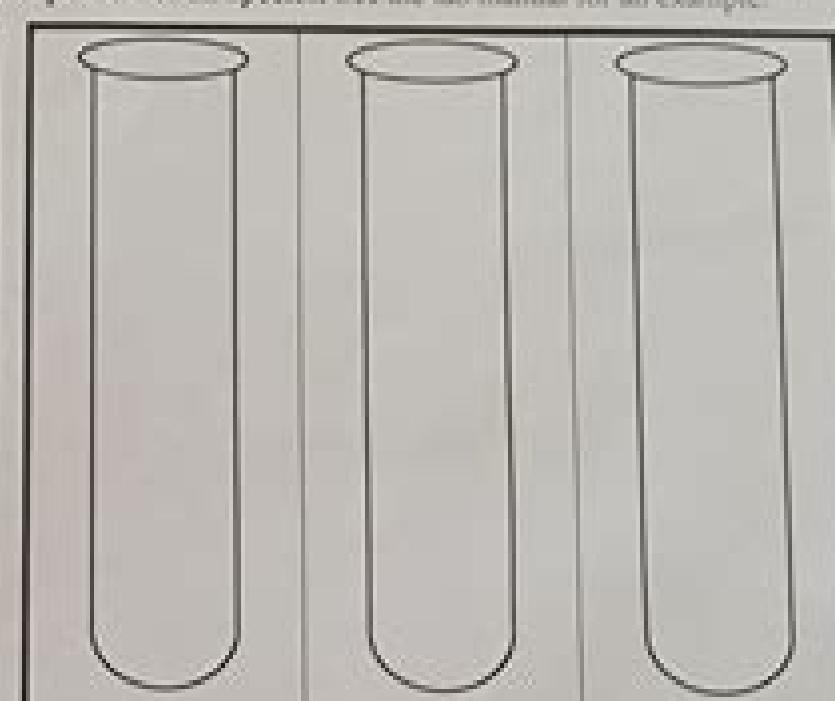


$\text{AgNO}_3$	$\text{Na}_2\text{SO}_4$	After mixing
-----------------	--------------------------	--------------

*Balanced chemical equation:*

*Complete ionic equation:*

*Net ionic equation:*



$\text{BaCl}_2$	$\text{NaBr}$	After mixing
-----------------	---------------	--------------

*Balanced chemical equation:*

*Complete ionic equation:*

*Net ionic equation:*

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**RJ Alexander**

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