

NAMING MOLECULAR COMPOUNDS

Name _____

Name the following covalent compounds.

1. CO_2 Carbon dioxide
2. CO Carbon Monoxide
3. SO_2 Sulfur Dioxide
4. SO_3 Sulfur trioxide
5. N_2O dinitrogen monoxide
6. NO nitrogen monoxide
7. N_2O_3 dinitrogen trioxide
8. NO_2 Nitrogen dioxide
9. N_2O_4 Dinitrogen tetroxide
10. N_2O_5 dinitrogen pentoxide
11. PCl_3 Phosphorus trichloride
12. PCl_5 Phosphorus Pentachloride
13. NH_3 Nitrogen trihydroxide
14. SCl_6 Sulfur hexachloride
15. P_2O_5 Diphosphorus pentoxide
16. CCl_4 Carbon tetrachloride
17. SiO_2 Silicon Dioxide
18. CS_2 Carbon disulfide
19. OF_2 Oxygen difluoride
20. PBr_3 phosphorus tribromide

1. Mono

2. di

3. tri

4. tetra

5. penta

6. hexa

7. hepta

8. octa

9. nona

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J Dewey



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