

## Chapter 9 - Nuclear Changes

**Alpha particle** – Is a positively charged atom that is released in the disintegration of radioactive elements and that consists of two protons and two neutrons

**Background Radiation** – The nuclear radiation that arises naturally from cosmic rays and from radioactive isotopes in the soil and air.

**Beta particle** – A charged electron emitted during certain types of radioactive decay, such as beta decay.

**Critical Mass** – The minimum mass of a fissionable isotope that provides the number of neutrons needed to sustain a chain reaction.

**Fission** – The process by which a nucleus splits into two or more fragments and releases neutrons and energy.

**Fusion** – The process in which light nuclei combine at extremely high temperatures. Forming heavier nuclei and releasing energy

**Gamma Ray** – the high-energy photon emitted by a nucleus during fission and radioactive decay

**Hal-life** the time required for half of a sample of a radioactive substance to disintegrate by radioactive decay or by natural processes.

**Nuclear Chain Reaction** – A continuous series of nuclear fission reactions.

**Nuclear Radiation** – The particles that are released from the nucleus during radioactive decay.

**Radioactivity** – The process by which an unstable nucleus emits one or more particles or energy in the form of electromagnetic radiation

**Radioactive tracer** – A radioactive material that is added to a substance so that its distribution can be detected later.

**Rem** – The quantity of ionizing radiation that does as much damage to human tissue as 1 roentgen of high-voltage x rays does.

# Key Terms Nuclear Changes Answers

**Diane E Papalia**



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